

ACROSS

1	A is the general name for
	esters of phosphoric acid.
3	is the chemical element that
	has the symbol P and atomic number
	15.
7	A bond is a group of strong
	covalent bonds between the
	phosphorus atom in a phosphate
	group and two other molecules over
	two ester bonds. Such bonds make up
	the backbone of DNA.
8	The cation is a positively
	charged polyatomic ion in which a
	phosphorus atom is bound to four
	hydrogens.
9	Alkyltriphenyl salts are
	widely used for the preparation of
	Wittig reagents for the Wittig reaction.
10	is a common
	organophosphorus compound in which
	a single phosphorus is bound to three
	phenyl groups.

DOWN

2	A is a phosphonic acid, its
	conjugate base, or the ester of a
	phosphonic acid.
4	Within organic chemistry, a
	acid contains an organic portion
	bound to a phosphorus which is singly
	bonded to two hydroxyl oxygens and
	doubly bonded to a third oxygen.
5	Phosphorus is widely used
	in the laboratory for the conversion of
	alcohols to alkyl bromides.
6	is the common name for
	phosphorus hydride, also known by
	the IUPAC name phosphane and,
	occasionally, phosphamine.